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A Novel Synthesis of 9,13-dicis Double Bonds Locked Retinoids

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Abstract: Synthesis of two retinoids in which the 9,13-dicis double bonds were locked in cycloalkene or thiophene was described. The key step was carbonylation of vinyl bromide 9 and 16 with carbon monoxide in the presence of Pd(PPh₃)₄. © 1997 Elsevier Science Ltd.

Retinoids are natural and synthetic analogs of Vitamin A. Because of their far ranging biological effects, retinoids have found clinical applications in dermatology, oncology and show promise in other diverse therapeutic areas including arthritis, dyslipidemias, and the prevention of HIV-induced lymphopenia. Cumulative evidence has indicated that retinoids may exert their functions by regulating gene expression mediated by two classes of nuclear receptors: the retinoic acid receptor family (RAR) 1 and the retinoid X receptor family (RXR)². The physiological hormones for the RAR and RXR are proposed to be all-trans -retinoic acid (RA) and 9-cis -retinoic acid (9-cis RA) respectively. However, 9-cis RA can bind to and transcriptionally activates the RAR as well 3. Unfortunately, use of the retinoids is associated with a number of significant side effects and therefore widespread clinical use of retinoids is severely limited. In view of the related, but clearly distinct, nature of these receptors, ligands which are selective for the RAR or RXR family would provide the capacity for independent control of the physiologic processes mediated by the RAR or RXR and further offer the possibility of improved therapeutic indices and reduced toxicity⁴. Recently, identification of 9,13-dicis-retinoic acid as a major plasma metabolite of 9-cis-RA has been reported⁵. In order to investigate the function of 9,13dicis RA and obtain the selective substances for RXR or RAR receptor, we were interested in designing and preparing compounds 1 and 2 in which the 9-cis and 13-cis double bond are locked in cyclohexene or thiophene and cyclopentene respectively.

Bromide 3 (Scheme 1), prepared in 85% yield by reaction of o-bromotoluene with 2,5-dichloro-2,5-dimethylhexane in the presence of aluminum chloride, was used as the starting material for the synthesis of 1. Bromide 3 was treated with *tert*-butyl lithium at -78°C, followed by zinc chloride at room temperature to give the organozinc compound, which is in turn reacted with ethyl 2-trifluoromethanesulfonyloxycyclohexene-1-ylcarboxylate 4 6 under palladium(0) catalysis 7 to give ester 5 in 85% yield. This is the first example of palladium-catalyzed cross-coupling of 2-carboethoxy enol triflate with an aryl zinc compound. Lithium aluminium hydride reduction of the ester 5 to the alcohol 6, followed by reaction with triphenylphosphine hydrogen bromide in dried methanol afforded the phosphonium salt 7 quantitatively. Wittig olefination of phosphonium salt 7 with 2-bromocyclopentenel-ylaldehyde 8 8 in dried dichloromethane in the presence of potassium *tert*-butoxide gave the single *trans* isomer vinyl bromide 9 in 94% yield. The *trans* isomer was determinated by the coupling constant of two vinyl protons (J_{H,H} =16.0 Hz). Carbonylation of the unstable vinyl bromide 9 with carbon monoxide in DMF under palladium(0) catalysis 9 at 85°C afforded methyl carboxylate 10 in 76% yield. The ester 10 was cleanly saponified to the 9,13-dicis-blocked acid 1¹⁰.

Scheme 1.

Reagents and Conditions: a) t-BuLi, -78°C, then ZnCl₂ room temperature, then Pd(PPh₃)₄ and 4; b) LiAlH₄; c) Ph₃P.HBr. MeOH; d) t-BuOK, CH₂Cl₂ then 8 e)CO, Pd(PPh₃)₄, MeOH, Et₃N, DMF, 85°C; f) aq.KOH, MeOH, reflux, then HCl

We then turned our attention to the synthesis of 2(Scheme 2). Our initial approach to the key intermediate 13 by the cross-coupling of the corresponding organozinc compound of aryl bromide 3 with methyl 3-bromo-2-thiophenecarboxylate 12¹¹ under palladium (0) or nickel(0) catalysis ¹² failed and the compound 12 was recovered. Therefore, an alternative route was adopted for preparation of 13. Aryl bromide 3 was reacted with t-butyllithium, followed by the addition of triisopropylborate and then hydrolysis to a boronic acid 11 in 75% yield. The Suzuki cross-coupling ¹³ of boronic acid 11 with 12 in the presence of Pd(PPh₃)₄ gave the ester 13 in 98% yield. Then the methodology used in Scheme 1 was extented to the preparation of 2. Thus, in an analogous manner, phosphonium salt 15 was prepared from 13. However, Wittig-olefination of phosphonium salt 15 with 8 afforded the mixture of vinyl bromide 16 in a ratio of trans:cis of 8:5 (as determined by ¹H NMR). The mixed vinyl bromide 16 could not be separated by column chromatography and used directly in carbonylation. The mixed ester 17 was hydrolyzed to give a solid which after recrystallization from absolute ethyl alcohol to afford the desired acid 2¹⁴.

Scheme 2.

Reagents and Conditions: a)t-BuLi,-78°C, then B(OPr-i)₃, R.T., HCl b) Pd(PPh₃)₄, NaHCO₃, ĎME, R.T.; c) LiAlH₄; d) Ph₃P.HBr, MeOH; e) t-BuOK, CH₂Cl₂; f)CO, Pd(PPh₃)₄, MeOH, Et₃N, DMF, 85°C; g) aq.KOH, MeOH, reflux, then HCl, Recrystallized from ethyl alcohol

The successful approach described herein, enabled us to synthesize a number of analogues related to 1 and 2. The structure-activity relationships of these derivatives and their ability to transactivate RAR or RXR selectively will be reported in a future publication.

References and Notes

- 1. (a)Petkovich, M.; Brand, N.J.; Kurst, A.; Chambon, P. *Nature* 1987, 330, 444. (b)Giguere, V.; Ong, E.S.; SEgui, P.; Evans, R.M. *Nature* 1987, 330, 624. (c) Benbrook, D.; Lerhardt, E.; Pfahl, M. *Nature* 1988, 333, 669.
- 2. (a)Hamada, K.; Gleason, S. L.; Levi, B. Z.; Hirschfeld, S.; Appella, E.; Ozato, K. Proc.Natl. Acad. Sci. U.S.A 1989, 86, 8289. (b)Mangelsdorf, D.J.; Ong, E.S.; Dyck, J.A.; Evans, R.M. Nature 1990, 345, 224. (c)Leid, M.; Kastner, P.; Lyons, R.; Nakshari, H.; Saunders, M.; Zacharewski, T.; Chen, J.Y.; Staub, A.; Garnier, J.M.; Mader, S.; Chambon, P. Cell 1992, 68, 377.
- 3. (a)Heyman, R.A.; Mangelsdorf, D.J.; Dyck, J.A.; Stein, R. B.; Eichelle, G.; Evans, R.M.; Thaller, C. Cell 1992, 68, 397. (b)Allenby, G.; Bocquel, M. T.; Sauders, M.; Kazmer, S.; Speck, T.; Rosenberger, M.; Lovey, A.; Kastner, P.; Grippo, J. F.; Chambon, P.; Levin, A. A. Proc. Natl. Acad. Sci. U.S.A. 1993, 90, 30. (c)Levin, A. A.; Sturzenbecker, L. J.; Kazmer, S.; Bosenberger, M.; Lovey, A.; Grippo, J. F. Nature 1992, 355, 359.
- 4. (a) Nadzan, A. M. Annual Reports in Medical Chemistry, 1995, Acdademic Press. Inc.; P. 119. (b) Dawson, M. I.; Jong, L.; Hobbs, P. D.; Cameron, J. F.; Chao, W. R.; Ptahl, M.; Lee, M. O.; Shroot, B.; Pfahl, M. J. Med. Chem. 1995, 38, 3368.
- 5. Tzimas, G.; Sass, J. O.; Wittfoht, W.; Elmazar, M. M. A.; Ehlers, K.; Nau, H. Drug Metab. Dispos. 1994, 22, 928.
- 6. Crisp, G. T.; Meyer, A. G. J. Org. Chem. 1992, 57, 6972.
- 7. Arcadi, A.; Byrini, A.; Cacchi, S.; Delmastro, M.; Marinelli, F.; Pietroni, B. Synlett. 1990, 47
- 8. Arnold, Z.; Holy, A. Collect. Czech. Chem. Commun. 1961, 26, 3059.
- 9. Cacchi, S.; Morera, E.; Ortar, G. Tetrahedron Lett. 1985, 26, 1109.
- 10. Compound 1: m.p. $264-264.5^{\circ}C$; ¹ H NMR(300 MHz, CDCl₃): δ : 1.22(s, 3H), 1.25(s, 3H), 1.26(s, 3H), 1.28(s, 3H), 1.72(s, 4H), 1.73-1.79(m, 6H), 2.08(s, 3H), 2.33-2.40(m, 6H), 2.67(t, J=7.0Hz, 2H), 6.27(d, J=16.0Hz, 1H), 6.89(s, 1H), 7.08(s, 1H), 7.37(d, J=16.0Hz, 1H)ppm; ¹³ C NMR(75 MHz,CDCl₃): δ : 19.18, 21.38, 22.78, 23.02, 25.11, 31.89, 32.09, 32.27, 33.20, 34.16, 35.37, 120.61, 126.70, 126.84, 127.68, 131.31, 132.13, 137.35, 139.49, 141.98, 143.27, 144.26, 156.24, 171.34 ppm; IR(KBr): γ_{max} : 3069, 3015, 2925, 2589, 1666, 1655, 1568, 1495, 1446, 1391, 1361, 1282, 1263, 1247, 978, 894, 766 cm⁻¹. UV(in CH₂Cl₂): 330 nm; MS (EI): 418(M⁺, 60), 385(84), 281(100), 195(55), 123(61), 111(75), 69(47), 57(49); HRMS: calcd for $C_{19}H_{15}O_{2}$ 418.2853, found 418.2834.
- 11. Tserng, K. Y.; Bauer, L. J. Org. Chem. 1975, 40, 172.
- 12. Negishi, E. I. Acc. Chem. Res. 1982, 15, 340.
- 13. Miyaura, N.; Yanagi, T. Suzuki, A. Synth. Commun. 1981, 11, 513.
- 14. Compound 2: m. p. 238-239°C; ¹H NMR(300 MHz,CDCl₃): δ : 1.27(s, δ H), 1.32(s, δ H), 1.70(s, δ H), 1.85(pent, δ H), 2.17(s, δ H), 2.62(t, δ H), 2.75(t, δ H), 2.75(t, δ H), 6.73(d, δ H), 6.73(d, δ H), 7.00(d, δ H), 7.08(s, δ H), 7.19(s, δ H), 7.27(d, δ H), 7.87(d, δ H), 7.87(d, δ H), 7.19(s, δ H), 7.27(d, δ H), 7.35(d, δ H), 7.35(d, δ H), 7.37(d, δ H), 7.35(d, δ H), 7.37(d, δ H)

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